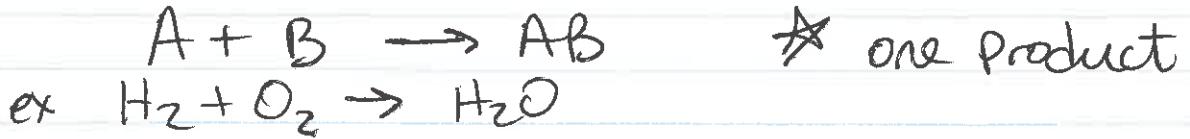


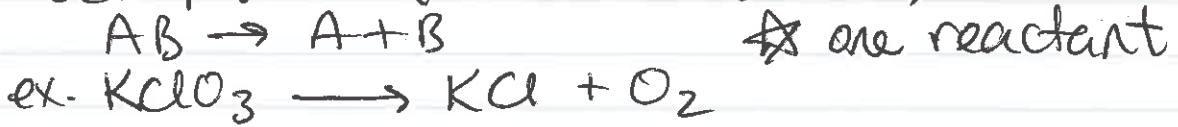
Types of Reactions

- 5 basic types of reactions.

① Synthesis (or combination)



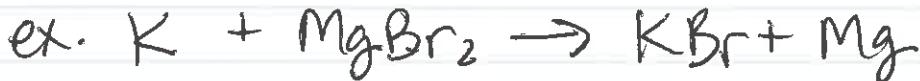
② Decomposition (reverse of a synthesis)



③ Single replacement (displacement)



\star (element + compound \rightarrow element + compound)



④ Double replacement (switching partners)



\star (compound + compound \rightarrow compound + compound)



⑤ Combustion



\star Product is always $CO_2 + H_2O$